

Simple Generation of a Reactive Glycosyl-lithium Derivative

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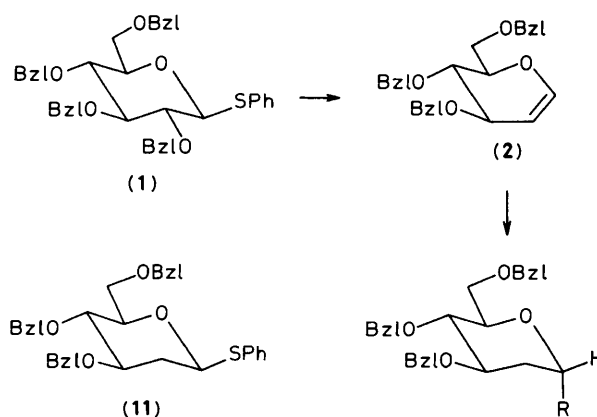
Two-step hydrolithiation of 3,4,6-tri-*O*-benzyl- β -D-glucal (hydrochlorination and lithium naphthalenide reductive lithiation) gives a reactive glycosyl-lithium derivative which is shown to be a precursor of *C*-glycosides.

A number of natural products regarded as *C*-glycopyranosyl derivatives have, because of their challenging structures,¹ caused intense interest in the field of carbohydrate chemistry. Most of the syntheses of *C*-glycosides developed so far² rely on the electrophilic character of the anomeric centre of a carbohydrate. Less conventional recent approaches derive either from the generation and trapping of a glucosyl radical³ or from the use of 1-deoxy-1-nitro sugars⁴ where the nitro group acts as an anion stabilizing substituent for the generation of a reactive anomeric anion in mild conditions.⁵ We report herein on the generation and reactivity of 3,4,6-tri-*O*-benzyl-2-deoxy- α -D-*arabino*-hexopyranosyl-lithium (**4**), the first example to-date of a *C*-glycopyranosyl-lithium reagent.

Treatment of phenyl 2,3,4,6-tetra-*O*-benzyl-1-thio- β -D-glucopyranoside (**1**)⁶ with 2 equiv. of lithium naphthalenide (LN) [tetrahydrofuran (THF), -78°C , 15 min] resulted in quantitative formation of 3,4,6-tri-*O*-benzyl- β -D-glucal (**2**).⁷ This result indicates that highly selective reductive lithiation⁸ occurred at the anomeric centre of (**1**), followed by a fast β -elimination.⁹

Hydrochlorination of (**2**) (toluene, HCl gas, 0°C , 10 min) gave a quantitative yield of 3,4,6-tri-*O*-benzyl-2-deoxy- α -D-*arabino*-hexopyranosyl chloride (**3**) which, upon reductive lithiation (2 equiv. LN, THF, -78°C , 3 min) and quenching with D_2O at -78°C , provided selectively the deuteriated derivative (**6**)† [80% from (**2**)], $[\alpha]_{\text{D}} +19^\circ$, most probably from the glycopyranosyl lithium (**4**).¹⁰ The axial addition of deuterium was shown by ^1H n.m.r. spectroscopy (90 MHz, CDCl_3): δ 3.92 (1 H, dd, $J_{1,2ax}$ 5.0, $J_{1,2eq}$ 2 Hz, 1-*Heq*).

Treatment of (**4**) with *p*-anisaldehyde gave a 3 : 1 diastereoisomeric mixture (**7**) (65%), from which the two pure diastereoisomers were separated on silica gel (toluene-ethyl acetate, 85 : 15 v/v): major product, $[\alpha]_{\text{D}} +16^\circ$; minor product, m.p. 65°C (from diethyl ether-hexane), $[\alpha]_{\text{D}} +9^\circ$. Oxidation (pyridinium chlorochromate, CH_2Cl_2 , room temperature, 5 h) of the mixture (**7**) gave a single product (**8**) (75%), m.p. $74\text{--}75^\circ\text{C}$ (from hexane), $[\alpha]_{\text{D}} +89^\circ$, ^1H n.m.r. (CDCl_3): δ 5.10 (1 H, dd, $J_{1,2ax}$ 6, $J_{1,2eq}$ 2 Hz, 1-H). Treatment of (**4**)‡ with



- (3) R = Cl
 (4) R = Li
 (5) R = H
 (6) R = D
 (7) R = *p*-MeOC₆H₄CH—
 (8) R = *p*-MeOC₆H₄CO—
 (9) R = Ph₂COH—
 (10) R = SPh

Bzl = CH₂Ph

benzophenone gave (**9**) (30%), $[\alpha]_{\text{D}} +15^\circ$, ^1H n.m.r. (CDCl_3): δ 1.68 (1 H, m, $J_{1,2}$ 4, $J_{2,3}$ 11, $J_{2eq,2ax}$ 13.5 Hz, 2-*Hax*). In this case, a large amount (30%) of (**5**) was also isolated. Similar results were obtained when phenyl 3,4,6-tri-*O*-benzyl-2-deoxy-1-thio- α -D-*arabino*-hexopyranoside (**10**), $[\alpha]_{\text{D}} +188^\circ$, or its β anomer (**11**), $[\alpha]_{\text{D}} -36^\circ$, were reductively lithiated¹¹ (2 equiv. LN, THF, -78°C , 45 min). Therefore the reductively generated glycosyl-lithium (**4**) appears to couple at -78°C so that electrophiles are introduced in the axial position.

In conclusion, the two-step hydrolithiation of tri-*O*-benzyl- β -D-glucal reverses the characteristic electrophilicity of the

† All new compounds gave satisfactory microanalytical and spectral data. Optical rotations were measured for solutions in chloroform at 20°C . The yields of the reactions have not been optimized.

‡ Reaction of (**4**) with chlorotrimethylsilane gave a single Si-glycoside (36%), $[\alpha]_{\text{D}} +15^\circ$, whose anomericism has not been ascertained.

§ (**10**) and (**11**) were prepared in four steps from (**3**) [(a) acetone- H_2O , Ag_2CO_3 ; (b) Ac_2O -pyridine; (c) PhSH, $\text{BF}_3\cdot\text{Et}_2\text{O}$, CH_2Cl_2 , room temperature, 15 min; (d) silica gel column (CH_2Cl_2 -hexane, 7 : 3, v/v)].

anomeric centre and thus forms a novel route to C-glycosides which complements existing procedures. The use of a phenyl thioglycoside as an excellent precursor to a glycal should also find useful applications.

We thank the Centre National de la Recherche Scientifique for financial support and Professor T. Cohen, University of Pittsburgh, for helpful advice.

Received, 28th December 1983, Com. 1661

References

- 1 For some typical examples, see F. J. McDonald, D. C. Campbell, D. J. Vanderah, F. J. Schmitz, D. M. Washecheck, J. E. Burks, and D. van der Helm, *J. Org. Chem.*, 1975, **40**, 665; D. T. Connor, R. C. Greenough, and M. von Strandtmann, *ibid.*, 1977, **42**, 3664; E. W. Colvin, S. Malchenko, R. A. Raphael, and J. S. Roberts, *J. Chem. Soc., Perkin Trans. I*, 1978, 658; G. R. Pettit, C. L. Herald, D. L. Doubek, D. L. Herald, E. Arnold, and J. Clardy, *J. Am. Chem. Soc.*, 1982, **104**, 6846; J. K. Cha, W. J. Christ, J. M. Finan, H. Fujioka, Y. Kishi, L. L. Klein, S. S. Ko, J. Leder, W. W. McWhorter, Jr., K.-P. Pfaff, M. Yonaga, D. Uemura, and Y. Hirata, *ibid.*, p. 7369.
- 2 S. Hanessian and A. C. Pernet, *Adv. Carbohydr. Chem. Biochem.*, 1976, **33**, 111; M. D. Lewis, J. K. Cha, and Y. Kishi, *J. Am. Chem. Soc.*, 1982, **104**, 4976, and references cited therein; R. M. Williams and A. O. Stewart, *Tetrahedron Lett.*, 1983, 2715, and references cited therein.
- 3 B. Giese and J. Dupuis, *Angew. Chem., Int. Ed. Engl.*, 1983, **22**, 622; R. M. Adlington, J. E. Baldwin, A. Basak, and R. P. Kozyrod, *J. Chem. Soc., Chem. Commun.*, 1983, 944.
- 4 B. Aebischer and A. Vasella, *Helv. Chim. Acta*, 1983, **66**, 789.
- 5 B. Aebischer, J. H. Bieri, R. Prewo, and A. Vasella, *Helv. Chim. Acta*, 1982, **65**, 2251.
- 6 R. J. Ferrier, R. W. Hay, and N. Vethaviasar, *Carbohydr. Res.*, 1973, **27**, 55.
- 7 For another synthesis see I. D. Blackburne, P. M. Fredericks, and R. D. Guthrie, *Aust. J. Chem.*, 1976, **29**, 381.
- 8 Selective reductive lithiation of a chloride or sulphide in the presence of benzyl ethers has been achieved: C. G. Screttas and M. Micha-Screttas, *J. Org. Chem.*, 1978, **43**, 1064.
- 9 Formation of glycals from an unstable C-1 anion has been reported: S. J. Eitelman and A. Jordaan, *J. Chem. Soc., Chem. Commun.*, 1977, 552; S. J. Eitelman, R. H. Hall, and A. Jordaan, *J. Chem. Soc., Perkin Trans. I*, 1978, 595; R. E. Ireland, C. S. Wilcox, and S. Thaisrivongs, *J. Org. Chem.*, 1978, **43**, 786; R. E. Ireland, C. S. Wilcox, S. Thaisrivongs, and N. R. Vanier, *Can. J. Chem.*, 1979, **57**, 1743.
- 10 Lithiomethyl methyl ether probably has a covalent structure in THF: U. Schöllkopf, *Angew. Chem., Int. Ed. Engl.*, 1970, **9**, 764, and references cited therein.
- 11 T. Cohen and J. R. Matz, *J. Am. Chem. Soc.*, 1980, **102**, 6900.